

Method Development and Validation of Bosentan by Uv-Spectrophotometric Method

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Abstract

A novel, safe, sensitive and easiest method of UV-spectro photometric method has been developed for the assay of Bosentan in its API and pharmaceutical dosage form. Method has been developed and validated using diluents like 0.1N sodium hydroxide at 261nm. All parameters were validated using ICH guidelines and Linearity was observed in the range 4-20 µg /ml for Bosentan ($r^2 = 0.999$) for the amount of drugs estimated by the proposed methods was in good agreement with the label claim.

Keywords: Spectrophotometric method, ICH, Bosentan, Sodium hydroxide.

Introduction

Ultraviolet-Visible spectrophotometry method

UV-Visible spectrophotometry is one of the most used methods in pharmaceutical analysis. This method involves measuring the amount of UV-Visible radiation absorbed by a drug in solution form. Instrument measures the ration of intensities of two lights in the UV-Visible region.

In qualitative analysis, drugs most are organic compounds can be identified by use of spectrophotometer, if any previous recorded data available for same compound. In quantitative analysis, quantity of compound is measured by studying the absorption radiation study of compound. UV-Visible spectrophotometric method is simple, rapid, specific and applicable to small quantities of compound [1].

Fundamental law that governs the quantitative spectrophotometric analysis is Beer-Lambert Law: when beam of light is passed through a transparent cell containing a solution of an absorbing substance, reduction of the intensity of light is mathematically calculated by using Beer-Lamberts law and it is expressed as follows,

$$A=abc$$

Where, A=absorbance, a=absorptivity, b=path length and c=concentration of solute in solution

Quantification of medical compound using UV-Visible spectrophotometer may carried out by preparing solution in transparent solvent and measuring it's absorbance at suitable wavelength [2-4]. The wavelength normally selected is wavelength of maximum absorption (λ_{max}). For assay of substance in multi component sample by spectrophotometer; the following methods are being used,

- Simultaneous estimation method
- Derivative spectrophotometric method
- Absorbance ratio method

Article Information

Article Type: Research

Article Number: JDDD106

Received Date: 13 December, 2018

Accepted Date: 21 December, 2018

Published Date: 28 December, 2018

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Citation: Pavani B, Harshavardhan B, Sindhu Priya G, Sharanya K, Kanchan (2018) Method Development and Validation of Bosentan by Uv-Spectrophotometric Method. J Drug Dev Del Vol: 1, Issu: 1 (34-38).

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- Difference spectrophotometry
- Solvent extraction method

Drug introduction

Bosentan: Bosentan is a dual endothelin receptor antagonist important in the treatment of pulmonary artery hypertension (PAH). It is licensed in the United States, the European Union and other countries by Actelion Pharmaceuticals for the management of PAH under the trade name Tracleer®. Bosentan is used to treat pulmonary hypertension by blocking the action of endothelin molecules that would otherwise promote narrowing of the blood vessels and lead to high blood pressure [5,6].

Endothelin-1 (ET-1) is a neurohormone, the effects of which are mediated by binding to ET_A and ET_B receptors in the endothelium and vascular smooth muscle. ET-1 concentrations are elevated in plasma and lung tissue of patients with pulmonary arterial hypertension, suggesting a pathogenic role for ET-1 in this disease [7]. Bosentan is a specific and competitive antagonist at endothelin receptor types ET_A and ET_B. Bosentan has a slightly higher affinity for ET_A receptors than for ET_B receptors.

Bosentan is eliminated by biliary excretion following metabolism in the liver. Terminal elimination half-life is about 5 hours in healthy adult subjects. Bosentan is metabolized in the liver by the cytochrome P450 enzymes CYP2C9 and CYP3A4 (and possibly CYP2C19), producing three metabolites, one of which, Ro 48-5033, is pharmacologically active and may contribute 10 to 20% to the total activity of the parent compound.

Methodology

Requirements

Table 1; Table 2

Choice of solvent

- With reference to Official compendia (E.P, B.P), it was found that Bosentan is sparingly soluble in water.

- Solubility check of the drug in various solvents like 0.1M Hydrochloric acid and 0.1MNaOH revealed the free solubility of drug in 0.1NSodiumhydroxide owing to the weak basic nature of drug.

- Hence 0.1N Sodium hydroxide was chosen as solvent for UV spectrophotometric analysis which gave distinct spectrum with Gaussian distribution and good absorbance.

Determination of λ_{max}

By trial and error, the λ_{max} of Bosentan in 0.1N sodium hydroxide by UV spectrophotometer was found to be 261.00 nm.

Method validation

Linearity and range: Linearity for the concentration range 4- 20 μ g/ ml was established by plotting concentrations on X- axis and corresponding absorbance on Y- axis. Statistical parameters like correlation coefficient (R^2), line equation including slope (m), y- intercept (C) were determined.

The specified range was derived from linearity studies by determining the difference between highest and lowest concentrations.

Precision

Intraday precision (repeatability): Repeatability of the developed method was assessed by 9 Determinations covering 3 concentrations each of 3 replicates. % RS was calculated for the results

Interday precision: Variations in the results for the developed method was assessed amidst 3 different days (n= 6). % RSD was calculated for the results obtained.

Limit of detection and limit of quantitation: LOD and LOQ were determined by instrumental methods based on the standard deviation of the response (blank sample) and slope of the calibration curve.

Accuracy: Accuracy of the method was confirmed by recovery studies from marketed formulation at three levels of standard addition from 50%, 100% and 150 % of label claim. The recovery studies were carried out in triplicate.

Preparation of 50% solution: 0.5ml of sample stock solution (5 μ g/ ml) and 0.25 ml of above standard stock solution II were pipetted out into a 10ml volumetric flask and diluted up to the mark with diluent.

Preparation of 100% solution: 0.5ml of sample stock solution (5 μ g/ ml) and 0.5 ml of above standard stock

Table 1: Requirements for Analysis of Bosentan by UV spectrophotometry.

Requirement	Manufacturer/Source
Calibrated UV visible spectrophotometer	Shimadzu UV – 1800
Calibrated electronic balance	Shimadzu
Bosentan	Cipla Ltd
Bosentan tablets	125mg
Volumetric flasks(10, 25, 1000 ml)	Class BBorosil
Pipettes (2,ml)	Class BBorosil
Beakers (50, 100, 250 ml)	Borosil
Simple distilled water	BhanuMini Quartz distiller

Table 2: Formulation for UV analysis.

Brand name	Bosentas
Active ingredient	Bosentan
Manufacturer	Cipla Ltd
Dosage form	Tablet- coated
Label claim	125 mg
Package	Blister – 10tablets per package
Expiry date	07/2020

Table 3: Linearity Profile by UV Spectrophotometry.

Concentration (μ g/ ml)	Absorbance at 261.00 nm
4	0.237
8	0.498
12	0.736
16	0.976
20	1.226

Table 4: Intra-day Precision Day- I by UV Spectrophotometry.

Conc (μ g/ ml)	Absorbance			Average	SD ^a	% RSD ^b
	Set 1	Set 2	Set 3			
4	0.236	0.238	0.235	0.236333	0.001528	0.646344
8	0.496	0.499	0.497	0.497333	0.001528	0.307143
12	0.735	0.737	0.736	0.736	0.001	0.13587

Table 5: Intra-day Precision Day- II by UV Spectrophotometry.

Conc (µg/ml)	Absorbance			Average	SD	% RSD
	Set 1	Set 2	Set 3			
4	0.237	0.236	0.235	0.236	0.001	0.423729
8	0.497	0.498	0.496	0.497	0.001	0.201207
12	0.737	0.735	0.736	0.736	0.001	0.13587

Table 6: Intra-day Precision Day- III by UV Spectrophotometry.

Conc (µg/ml)	Absorbance			Average	SD	% RSD
	Set 1	Set 2	Set 3			
4	0.236	0.235	0.238	0.2363	0.001528	0.646344
8	0.497	0.498	0.499	0.498	0.001	0.200803
12	0.735	0.736	0.737	0.736	0.001	0.13587

^a= Standard Deviation, ^b= Percentage Relative Standard Deviation

Table 7: Inter-day Precision by UV Spectrophotometry.

Conc (µg/ml)	Absorbance						Average	SD	% RSD
	Set 1	Set 2	Set 3	Set 4	Set 5	Set 6			
4	0.238	0.235	0.236	0.237	0.235	0.236	0.23616	0.001169	0.495009
8	0.499	0.497	0.498	0.497	0.498	0.496	0.4975	0.001049	0.210816
12	0.737	0.736	0.735	0.737	0.736	0.735	0.736	0.000894	0.121525

Table 8: Recovery from Formulation (Bosentan tablets) by UV spectrophotometry.

Bosentan in dosage form (µg ml ⁻¹)	% Pure Bosentan added	Pure Bosentan Added (µg ml ⁻¹)	Bosentan Recovered % ± %RSD*
5	50%	2.5	101.02±1.17
5	100%	5	101.90±1.12
5	150%	7.5	102.00±0.17

*Average of 3 experiments

Table 9: Assay of formulation (Bosentan 125mg tablets) by UV spectrophotometry.

Formulation	Absorbance	Label claim	Amount found	% Assay ± SD*
Bosentan	0.741	125 mg	11.934	99.45%
	0.735			
	0.731			

solution II were pipetted out into a 10ml volumetric flask and diluted up to the mark with diluent.

Preparation of 150% solution: 0.5 ml of sample stock solution (5µg/ml) and 0.75ml of above standard stock solution II were pipetted out into a 10ml volumetric flask and diluted up to the mark with diluent.

Applicability of the developed validated method by ultraviolet spectrophotometry

Assay of formulation: 20 tablets were weighed accurately and average weight of tablet was noted that constitutes 125 mg Bosentan and was finely powdered.

The tablet powder equivalent to 0.0175mg of Bosentan was accurately weighed and transferred to 10 ml volumetric flask and dissolved in about 5 ml of the solvent (0.1 N Sodium hydroxide).

It was then vortexed for 30 minutes to enhance maximum extraction of the active pharmaceutical ingredient from the dosage form and filtered through Whatmann No 1 filter paper to remove insoluble excipients to the maximum extent.

It was then made up to the volume with the same solvent. This constitutes 10µg/ml of Bosentan.

From the stock solution, aliquot corresponding to medium concentration of standard curve (12µg/ml) was prepared and made up to the mark with the solvent.

The absorbance was noted and the corresponding concentration was then determined from the standard calibration curve.

Results and Discussion

Method development and validation of Bosentan drug by UV

Determination of λ_{max} of Bosentan: Figure 1

Method validation:

Linearity: The results obtained were within the range from the linearity set, shows good correlation coefficient value (0.9998). Figure 2 and 3; Table 3

Precision

Intra-day precision (Repeatability): Table 4-6

Inter-day precision: The developed method was found to be precise as the % RSD of the results within and amidst 3 days was within limits (< 2.0). Table 7

Limit of detection: LOD = 3.3 δ/ m

$$= 3.3 \times 0.001528 / 0.0614$$

$$= 0.082124 \mu\text{g}$$

Limit of quantitation: LOQ = 10 δ/ m

$$= 10 \times 0.001528 / 0.0614$$

$$= 0.24886 \mu\text{g}$$

The obtained results for Limit of Detection (LOD) and Limit of Quantification (LOQ) were satisfactory.

Accuracy: *Average of 3 experiments

Acceptance Criteria: The % Recovery for each level should be between 98.0 and 102.0%

The developed method was found to be accurate since % recovery for each level was within limits and RSD less than 2.0. Figure 4; Table 8

Applicability of the developed validated method by ultraviolet spectrophotometry

Assay of formulation: Acceptance criteria: 95- 105% w/v

The assay results were decorous in conjunction with acceptance criteria. Table 9

Discussion

The proposed methods were validated. The accuracy of the methods was assessed by recovery studies at three different levels. Recovery experiments indicated the absence of interference from commonly encountered pharmaceutical additives. The method was found to be precise as indicated by the repeatability analysis, showing %RSD less than 2. All statistical data proves validity of the methods and can be used for routine analysis of pharmaceutical dosage form.

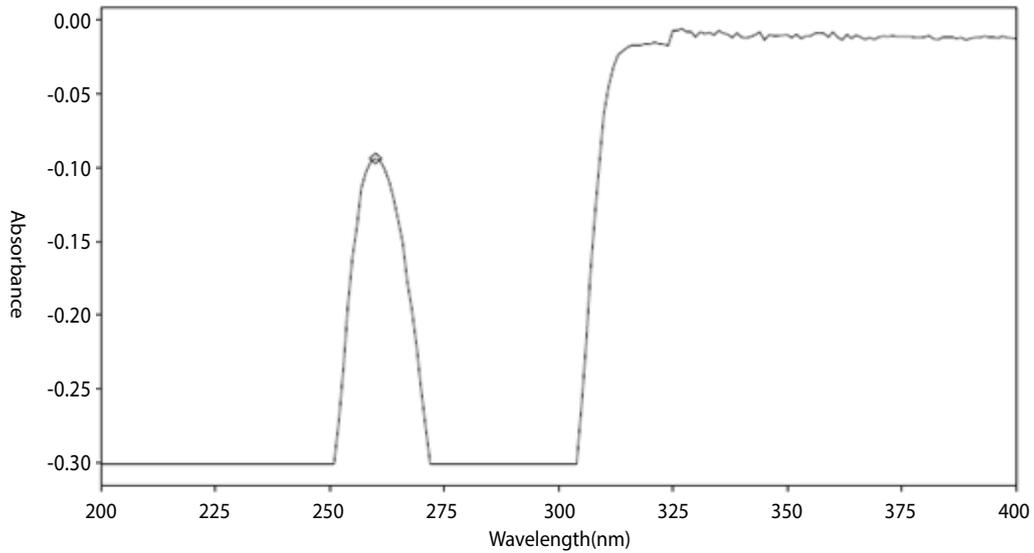


Figure 1: λ_{max} of Bosentan.

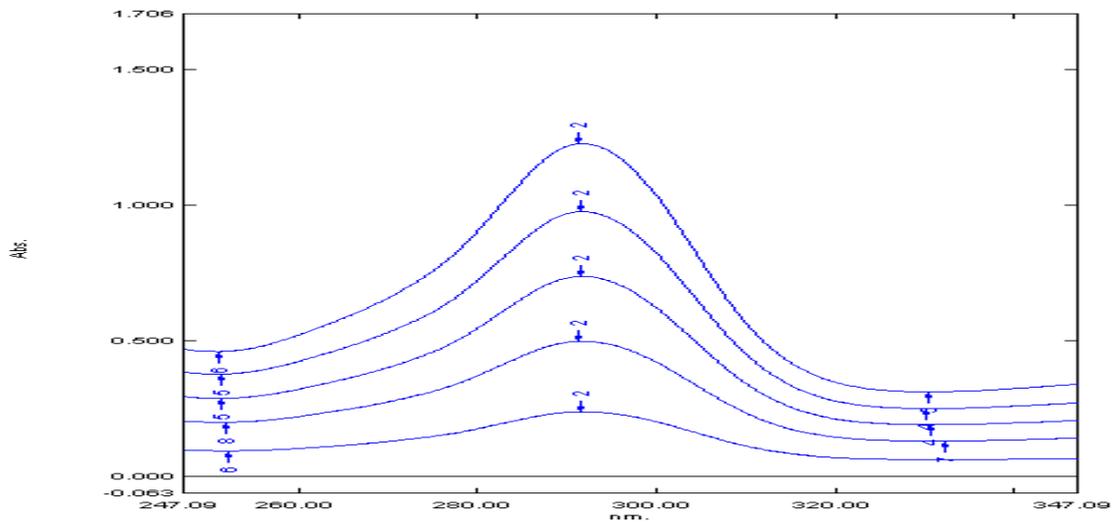


Figure 2: Overlay Spectra of Bosentan (4-20 µg/ ml).

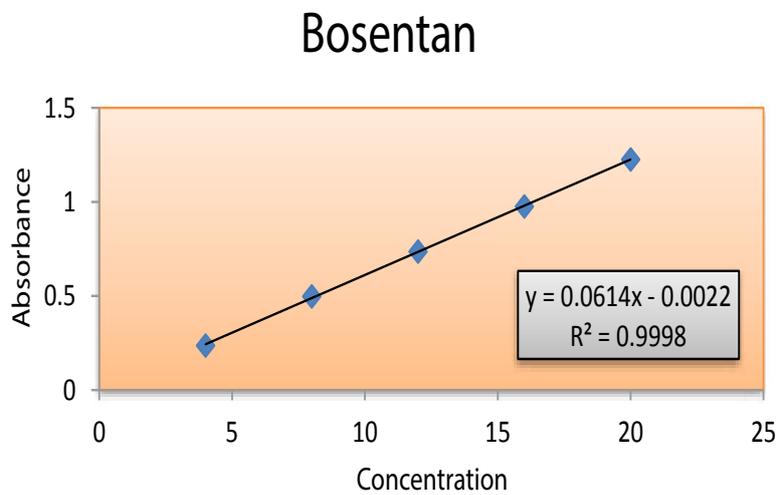


Figure 3: Calibration Curve for the Linearity Set at 261.00 nm by UV Spectrophotometry.

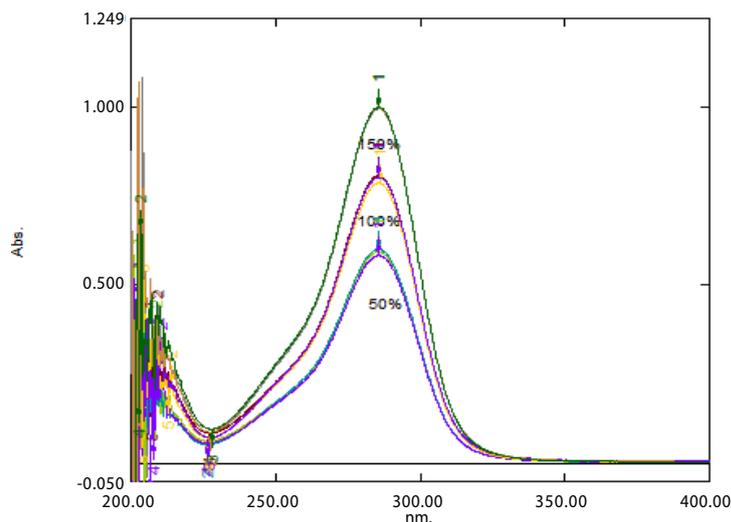


Figure 4: Overlay Accuracy Spectra of BOSENTAN by UV Spectrophotometry.

Conclusion

From the above experimental results and parameters it was concluded that, this newly developed method for the estimation of Bosentan was found to be simple, precise and accurate; this method more acceptable and cost effective and it can be effectively applied for routine analysis in research institutions, quality control department in meant in industries, approved testing laboratories studies in near future.

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