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Study on the In-vitro Release Kinetics of Olmesartan-loaded Optimized Solid Lipid Nanoparticles

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Abstract

Drugs with low aqueous solubility not only give low oral bioavailability but provide high inter-and intra-subject variability. Solid lipid nanoparticles (SLN) have gained much interest as potential drug carriers for lipophilic drugs. Olmesartan medoxomil has very poor aqueous solubility and belongs to Class II drugs under Biopharmaceutical Classification Systems. The objective of the present study is to determine release kinetics of Olmesartan medoxomil- loaded solid lipid nanoparticles. Olmesartan medoxomil-loaded solid lipid nanoparticles was prepared by hot homogenization and ultra-sonication method. Particle size, polydispersity index, shape and surface morphology were used to characterize the formulations. The degree of encapsulation of the drug in the formulations was estimated by determining the drug entrapment efficiency. The *in-vitro* drug release from the formulations was studied using dialysis membrane. The drug release kinetics of the formulation was analyzed using different kinetics models. The results show that the formulations are spherical in shape with smooth surface and possess particle size range of 122.8-135.0 nm, polydispersity index range of 0.208-0.239. The entrapment efficiency was in range of 94.5-96.8%. Fourier transform infrared spectroscopy (FTIR) and differential scanning colorimetry (DSC) test results are considered to be satisfactory. The in-vitro drug release study demonstrated that drugloaded formulations possessed controlled drug release characteristics. The kinetics release results suggest the possible mechanism of action for the drug release might be diffusion of the drug from the formulation. Zero-order drug release kinetics model out of the models investigated, best fitted the in-vitro release data.

Keywords: Olmesartan, Nanoparticles, Nanoemulsions, Liposomes.

Introduction

A number of lipid-based delivery systems for enhancing the bioavailability of lipophilic drugs have been reported. Such systems include microemulsion [1-3], nanoemulsions [4-6], liposomes [7-10], self-emulsifying drug delivery systems [11-13]. Solid lipid nanoparticles are colloidal carrier systems that have advantages over these particulate drug carriers [14-16]. The solid lipid nanoparticles are composed of physiological lipid, dispersed in water or in an aqueous surfactant solution. The solid lipid core may contain triglycerides, glyceride mixtures, or waxes that are solid at both room temperature and human body temperature [17-19]. Due to their good biocompatibility, low toxicity, physically stability and better delivery of lipophilic drugs these solid lipid nanoparticles formulations have been developed for various routes of applications namely dermal, ocular, oral, parenteral, pulmonary and rectal. These formulations can be used to improve the bioavailability

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of drugs [20, 21], obtain sustained release of lipophilic drugs [22-24]. Olmesartan medoxomil (Figure 1-3) chemically 2.3-dihydroxy-2-butenyl 4-[1-hvdroxy-1methylethy]-2-propyl-1-[p(o-1H-tetrazol-5-ylphenyl) benzyl] imida-zole-5-carboxylate, cyclic 2,3-carbonate is a selective AT1 subtype angiotensin-II receptor antagonist. It acts by lowering blood pressure through arterial vasodilatation and reduced sodium retention [25]. Clinically, it used in the treatment of hypertension [26]. Olmesartan medoxomil is available only as tablets in doses of 5 mg, 20 mg and 40 mg respectively [27]. It is practically insoluble in water (<7.75 µg/ml) and rapidly absorbed from the gastrointestinal tract with peak plasma concentration of olmesartan (metabolite) occurring 1-3 hafter administration. Olmesartan medoxomil is converted during absorption to olmesartan, which is subsequently excreted without further metabolism [28]. The absolute bioavailability of olmesartan from olmesartan medoxomil tablets is 28.6% with an elimination half-life of 10-15 hr [29]. Literature survey has shown that some methods [30-33] have been used to improve the bioavailability of olmesartan medoxomil, however, as solid lipid nanoparticles delivery system has not been explored for olmesartan medoxomil, the drug based on its physicochemical and biopharmaceutical properties was selected as a good candidate for solid lipid nanoparticle formulation. In-vitro dissolution studies/drug releases are very vital for the development of new pharmaceutical dosage forms and serves as an alternative to in-vivo bioavailability determination [34]. Good correlations have been established between in-vitro dissolution/drug release and in-vivo absorption [35]. A number of mathematical models [36] have been exploited in determining the drug release kinetics. These predictive mathematical models describe the relationship between in-vitro property of dosage form and a relevant in-vivo response. Therefore, the objective of the present study was to investigate olmesartan medoxomil-loaded solid lipid nanoparticles release kinetics using predictive mathematical models while envisaging that the results obtained will support the use solid lipid nanoparticles formulation as a potential delivery system for olmesartan medoxomil.

Materials and Methods

Material

Olmesartan medoxomil (Sankyo Pharma Inc., USA), glyceryl mono stearate (Sigma-Aldrich, USA), soya lecithin (Himedia Ltd, India), polysorbate 80 (Sigma-Aldrich, USA). All other chemicals were of analytical reagent grade

Preparation of olmesartan medoxomil loaded-solid lipid nanoparticles

Olmesartan medoxomil, glyceryl mono stearate, and soya lecithin were dissolved in a beaker containing 10 mL mixture of chloroform and methanol (4:1). Organic solvents were completely evaporated on a water bath. The lipid layer having the embedded drug was melted at temperature 5°C above melting point of the lipid. An aqueous phase containing polysorbate 80 heated to the same temperature as the oil phase was slowly added to the oil phase while homogenization was carried out at 12,000 rpm using Ultra Turrex T25 homogenizer (IKA labotechnik) for 5 min. The coarse hot oil in water emulsion was ultrasonicated (12T-probe) using a Vibra cell (Sonics, USA) for 20 min. Olmesartan medoxomil loaded-solid lipid nanoparticles formulation was got by allowing hot formulation to cool to room temperature.

Characterization of solid lipid nanoparticles formulation

Measurement of size and polydispersity index

The mean particle size and polydispersity index of the nanoparticles preparations were determined using a zetasizer (DTS Version 4.10, Malvern instruments, UK). Samples were appropriately diluted with deionised water to allow the light scattering intensity to be within the instrument's sensitivity range.

Fourier transformed infrared (FTIR) spectroscopy

The samples were analyzed by Fourier transform infrared (FT-IR) spectroscopy in the range of 400 to 4,000 cm–1. The infrared spectral analysis of pure olmesartan medoxomil, glyceryl mono stearate, soya lecithin, physical mixture of drug plus excipients and olmesartan medoxomil loaded-solid lipid nanoparticles formulations were carried out. The vibration frequencies of the spectra peaks produced by the pure drug, physical mixture and solid lipid nanoparticles formulation were compared.

Scanning electron microscopy

Shape and surface morphology of the solid lipid nanoparticles formulations were analyzed using scanning electron microscopy (SEM). The samples were mounted on alumina stubs using double adhesive tape, coated with gold in HUS-5GB vaccum evaporator. The analysis was done using Hitachi S-3000N SEM at an acceleration voltage of 10KV and a magnification of 5000×.

Differential scanning colorimetry

The test was done on a calibrated instrument under nitrogen purge (20 mL/min) at a heating rate of 100C/min and temperature range of 25–2000°C using Netzsch DSC

200PC (Netzsche, Selb, Germany). The sample was weighed into standard aluminum pan while using the empty pan as reference.

Total drug content determination

A 2 ml of each solid lipid nanoparticles formulation was diluted to 10 ml using ethanol. The absorbance of the drug in all dilutions, were measured at 258 nm. The total drug content in each formulation was obtained from calibration curve.

Determination of entrapment efficiency (EE)

The entrapment efficiency was done by obtaining the concentration of free drug in the aqueous phase of solid lipid nanoparticles formulation following centrifugation of 2 ml of each formulation. Centrifuging of the formulation was done at high speed (15000 rpm) for 30 min at room temperature using Remi cooling centrifuge (Mumbai, India). Absorbance of the aqueous solution was taken at a wavelength of 258 nm. The entrapment efficiency was calculated as follows:

% EE = Total drug content-free drug content
______×100

Total drug content

Stability Study

Stability testing was done on olmesartan-loaded solid lipid nanoparticles formulations at room and refrigerator temperatures for 0, 30 and 60 days. Particle size, polydispersity index and entrapment efficiency determinations were parameters used to follow the stability

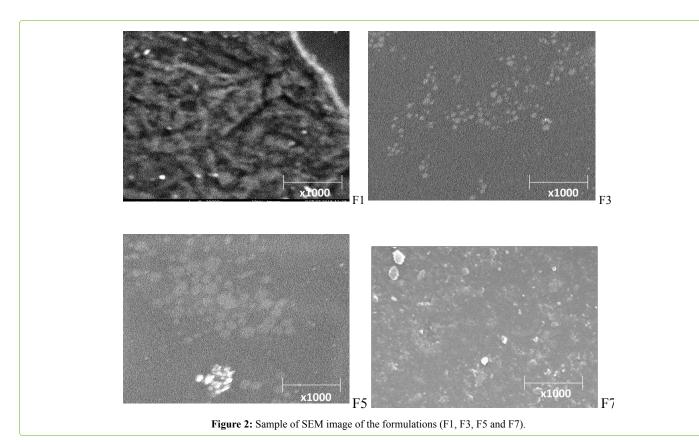
study. Study was done in triplicate. Statistical analysis of the data was performed using the Student's *t*-test and significance of the study was chosen at p<0.05.

In-vitro drug release

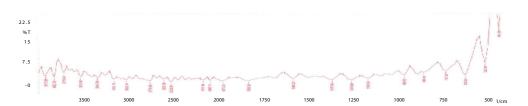
Dialysis bag was used for the in-vitro drug release determination. The bag was soaked in distilled water for 12 h prior to been used. A 5 ml aliquot of solid lipid nanoparticles formulation devoid of free drug or olmesartan medoxomil suspension was placed into dialysis bag and tied at both ends. The dialysis bag was immersed in a receptor compartment containing 50ml of saline phosphate buffer solution (PBS) pH 7.4. The receptor compartment was placed on a magnetic stirrer and stirred at 37±1°C using a magnetic bar. At various time intervals (5, 10, 20, 30, 40, 50, 60, 90, 120, 180, 240, 300 min up to 12 h) a 5 ml of aliquot was withdrawn, filtered using Whatsman filter paper. An equal volume of fresh medium was added to the receptor compartment. The filtered samples were analyzed for drug contents by UV-Visible spectrophotometry at 258 nm. Plot of percent of drug released versus time was carried out.

Drug release kinetics

In-vitro drug release data were fitted into various kinetic models such as zero order (cumulative amount of drug released versus time), first order (log cumulative percentage of drug remaining versus time), Higuchi model (cumulative percentage of drug released versus square root of time) Hixson Crowell model (cubic root of initial drug concentration minus cubic root of drug concentration at a given time versus time, $Qo^{1/3}$ - $qt^{1/3}$) vs t.) and Korsmeyer-peppas model (logarithm of fraction of drug released versus logarithm of time, $log(Qt/Q\infty)$ vs log t). The best-fit model



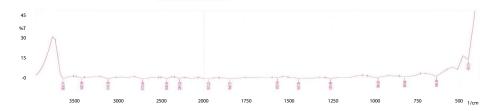
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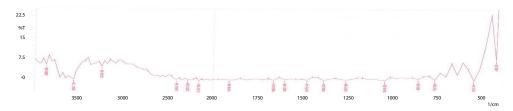
FT-IR Scan of Pure OMS



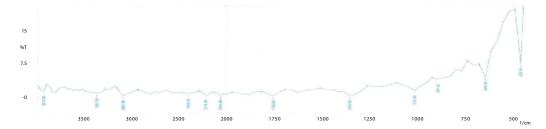
FT-IR Scan of GMS



FT-IR Scan of Soy lecithin



FT-IR Scan of polysorbate 80



FT-IR Scan of physical mixture

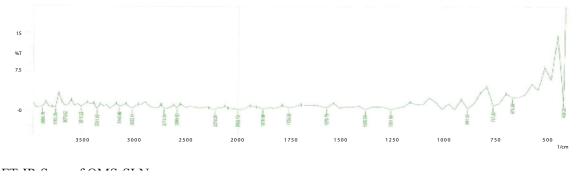


Figure 3: Fourier Transform Infrared spectroscopy of Olmesartan Medoxomil glyceryl monosterate soy lecithin, polysorbate 80, physical mixture and solid lipid nanoparticles formulation.

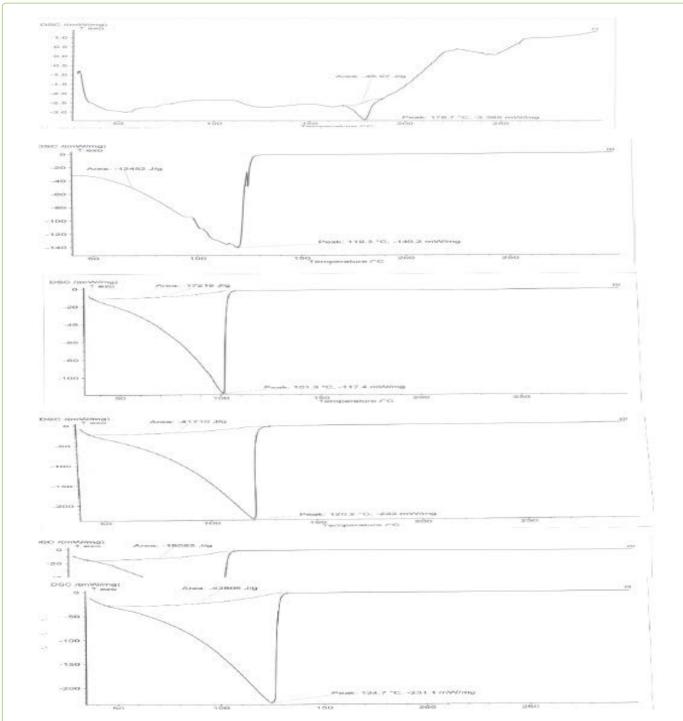


Figure 4: Differential scanning calorimetry of Olmesartan Medoxomil, glyceryl monosterate, soy lecithin, polysorbate 80, physixal mixture and solid lipid nanoparticles formulation.

was chosen based on the highest correlation coefficient obtained from the regression analysis.

Results and Discussion

Preparation of olmesartan medoxomil loaded-solid nanoparticles

The homogenization followed by ultrasonication technique used to prepare the solid lipid nanoparticles formulations have been reported to be simple, reliable and reproducible method for preparation of solid lipid

nanoparticles [37]. The solvent mixture of chloroform and methanol (4:1) was found to be very effective in homogenously dispersing olmesartan medoxomil in the lipid phase. The composition of the formulations, are given in Table 1.

Particle size distribution and polydispersity index of formulations

The particles of size range of 122.8-135.0 nm, polydispersity index range of 0.208-0.239 were obtained for olmesartan medoxomil-loaded solid lipid nanoparticles

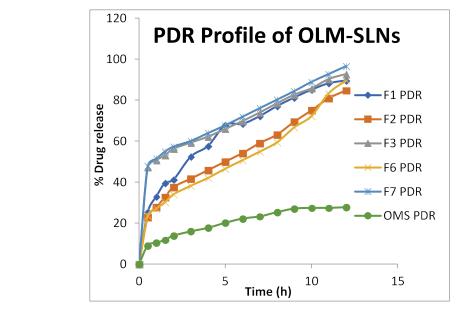


Figure 5: Percentage drug release graph of OLM-SLNs formulations.

formulations (Table 2). The particle size results suggest that the formulations have the potential to give high drug release from the nanoparticles matrix, good gastrointestinal uptake and *in-vivo* distribution of the drug. Previous report has shown that particle size less than 300 nm are advisable for the intestinal transport [38]. The polydispersity index results also substantiated the optimum size distribution of the nanoparticles.

It has been reported [39] that a polydispersity index value less than 0.3 is often accepted as optimum value.

Scanning electron microscopy:

The test showed that most of the particles were smooth and fairly spherical in shape. It was also noted that the particle sizes of the formulations were in the nanometric range.

Fourier transformed infrared (FTIR) spectroscopy

The infrared spectroscopic studies were done to determine possible drug-lipid interactions. The IR spectrum of pure olmesartan medoxomil showed sharp characteristic peaks as given in Table 3. The characteristic peaks appear in the physical mixture indicating no modification or interaction between drug and the excipients. However, these characteristic peaks were not found in the solid lipid nanoparticles formulations suggesting that the drug might have been molecularly dispersed within the lipid matrix.

Differential scanning colorimetry analysis

The DSC thermograms of olmesartan medoxomil and drugloaded solid lipid nanoparticles formulation (F3) are shown in (Figure 4,5). The thermogram of olmesartan medoxomil shows endothermic peak at 178.7°C. It was observed that the melting endotherm of olmesartan medoxomil was shifted to lower temperatures in formulations. For example it was shifted to $110.4^{\circ C}$ for F3 in the thermograms of drug-loaded solid lipid nanoparticles. The observation tends to suggest that olmesartan medoxomil was not in crystalline state but

completely solubilized in the formulation.

Entrapment efficiency

The results of the entrapment efficiency of olmesartan medoxomil-loaded solid lipid nanoparticles formulations are listed in Table 2. Among the formulations, highest entrapment efficiency (96.8 \pm 0.03%) was observed with formulation of SLN F7, whereas formulation SLN f5 showed lowest entrapment efficiency (79.4 \pm 0.05%). As the lipid concentration increased there was a decline in entrapment efficiency of the formulation.

Stability study

The results are in Table 4 shows that there was no significant difference (P<0.05) in particle size and polydispersity index values. Although the zeta potential of the formulations is not reported in the present study, the results of the stability study suggest that the zeta potentials of the formulations are within the acceptable range. Zeta potential is a key factor for evaluation of the stability of colloidal dispersions.

In-vitro drug release

To evaluate the release rate of olmesartan medoxomil-loaded solid lipid nanoparticles formulations, the release of olmesartan medoxomil from the lipid particles was investigated for 12 h in saline phosphate buffer pH 7.4. All formulations showed sustained release of olmesartan medoxomil, and the cumulative percent drug release of olmesartan medoxomil loaded SLNs after 12hrs was found to be 96.5%, 92.7%, 89.6 % etc while that of aqueous drug suspension was 27.8 %. From the results, it was observed that, all the formulations showed higher release from all the formulations than the aqueous drug suspension, implying improved bioavailability. The results are shown in Table 5.

Drug release kinetics

The release data of olmesartan medoxomil-loaded solid

Table 1: Composition of Olmesartan Medoxomil solid lipid nanoparticles formulations.

Ingredients	Formulations							
ringredients	F1	F2	F3	F4	F5	F6	F7	
Olmesartan medoxomil (% w/w)	0.038	0.038	0.038	0.037	0.037	0.037	0.036	
Glyceryl monosterate (% w/w)	1.89	1.89	1.89	2.80	3.70	1.85	1.82	
Soy lecithin (% w/w)	1.89	1.89	1.89	1.87	1.85	2.78	3.64	
Polysorbate 80 (% w/w)	1.89	1.89	1.89	1.87	1.85	2.78	3.64	
Water (% w/w)	94.30	94.30	94.30	93.42	92.56	92.56	90.88	

Table 2: Particle size, polydispersity index, percent entrapment efficiency and percent drug content.

Formulation	Particle size (nm)	Polydispersity	Entrapment efficiency (%)	Drug content (%)
F1	137.6 ± 2.33	0.239 ± 0.12	94.50 ± 0.05	96.8 ± 0.05
F2	145.1 ± 2.38	0.225 ± 0.18	95.20 ± 0.07	97.9 ± 0.06
F3	135.0 ± 2.23	0.206 ± 0.03	96.00 ± 0.09	98.5 ± 0.05
F4	199.8 ± 2.13	0.564 ± 0.16	85.00 ± 0.10	92.4 ± 0.06
F5	242.1 ± 4.27	0.463 ± 0.18	79.04 ± 0.12	90.2 ± 0.08
F6	144.9 ± 5.47	0.210 ± 0.11	95.50 ± 0.06	97.7 ± 0.07
F7	122.8 ± 2.30	0.208 ± 0.13	96.75 ± 0.03	98.9 ± 0.03

Table 3: FTIR Spectra.

Ingredient	Functional groups with wave numbers (cm ⁻¹)								
	C-H (aromatic)	C-N (stretch)	C=C (aromatic)	C=O (carboxylic)	O-H (str.)	O-H (bend)	C-O-C (ring, str.)		
Olmesartan medoxomil	2959.96	1169.34	1586.22	1733.26	3531.66	absent	absent		
Glyceryl monosterate	2930.32	1184.78	1647.98	1725.18	3553.74	3176.54	1045.82		
Soy lecithin	2913.34	1261.98	absent	1783.24	3639.74	3520.44	absent		
Physical mixture	2850.23	absent	absent	1756.06	3547.10	3361.82	1014.94		
Solid lipid Nanopaticles	2913.34	absent	1756.06	1733.26	3763.26	3670.62	absent		

Table 4: Stability studies of Olmesartan Medoxomil loaded SLN formulations (room and refrigerator temperature).

Day 1 (4 °C and 25 °C)			Day 30 (4 °C and 25 °C)			Day 60 (4 °C and 25 °C)			
Form Code	particle size (nm) mean ± S	PDI mean ± SD	EE (%)	particle size (nm)	PDI mean ± SD	EE (%)	Particle size (nm)	PDI mean ± SD	EE(%)
F1	136.6 ± 2.33	0.233 ± 0.12	94.50 ± 0.05	138.8 ± 2.23	0.233 ± 0.15	95.13 ± 0.11	136.6 ± 2.33	0.232 ± 0.22	94.14 ± 0.24
F2	145.1 ± 2.38	0.225 ± 0.18	95.20 ± 0.10	146.0 ± 4.58	0.220 ± 0.17	96.27 ± 0.04	144.2 ± 2.34	0.224 ± 0.15	95.90 ± 0.20
F3	135.0 ± 2.23	0.206 ± 0.03	96.00 ± 0.09	134.0 ± 3.12	0.204 ± 0.13	97.32 ± 0.08	135.3 ± 3.00	0.205 ± 0.14	96.77 ± 0.21
F4	199.8 ± 2.13	0.564 ± 0.16	80.00 ± 0.15	200.6 ± 2.15	0.562 ± 0.14	80.50 ± 0.04	198.1 ± 2.10	0.562 ± 0.19	81.60 ± 0.19
F5	242.1 ± 2.27	0.463 ± 0.18	86.04 ± 0.13	243.3 ± 2.24	0.462 ± 0.12	85.00 ± 0.13	243.3 ± 2.12	0.440 ± 0.16	86.55 ± 0.15
F6	144.9 ± 2.47	0.210 ± 0.11	95.50 ± 0.06	144.2 ± 3.27	0.211 ± 0.09	94.05 ± 0.12	143.0 ± 2.25	0.212 ± 0.08	95.70 ± 0.17
F7	122.8 ± 2.30	0.208 ± 0.13	96.75 ± 0.03	124.4 ± 2.32	0.206 ± 0.11	97.19 ± 0.04	121.1 ± 2.12	0.205 ± 0.18	96.56 ± 0.14

 Table 5: In vitro drug release profile.

Time (h)	Time (h) % Cumulative drug release						
	Olmesartan medoxomil	F1	F2	F3	F6	F7	
0	0	0	0	0	0	0	
0.5	09.03 ± 0.12	25.35 ± 0.63	22.92 ± 0.00	47.22 ± 0.00	23.61 ± 0.46	47.91 ± 0.00	
1.0	10.41 ± 0.00	32.99 ± 0.01	27.78 ± 0.67	50.69 ± 0.12	27.08 ± 0.46	51.38 ± 0.00	
1.5	11.81 ± 0.40	39.58 ± 0.58	32.63 ± 0.57	53.13 ± 0.63	30.21 ± 0.13	54.86 ± 0.53	
2.0	13.88 ± 0.06	46.18 ± 0.58	37.58 ± 0.00	56.25 ± 0.00	34.03 ± 0.53	57.29 ± 0.00	
3.0	15.97 ± 0.69	52.43 ± 0.00	41.67 ± 0.00	59.38 ± 0.69	38.19 ± 0.52	60.07 ± 0.00	
4.0	17.71 ± 0.00	57.63 ± 0.58	45.83 ± 0.52	62.15 ± 0.52	42.01 ± 0.00	63.89 ± 0.49	
5.0	20.14 ± 0.20	62.84 ± 0.58	50.00 ± 0.52	65.97 ± 0.00	46.53 ± 0.00	67.71 ± 0.00	
6.0	22.22 ± 0.00	68.40 ± 0.58	54.17 ± 0.00	69.79 ± 0.64	50.69 ± 0.00	71.88 ± 0.00	
7.0	23.26 ± 0.05	72.22 ± 0.00	59.03 ± 0.00	73.96 ± 0.52	54.86 ± 0.00	76.04 ± 0.00	
8.0	25.34 ± 0.00	77.08 ± 0.58	63.19 ± 0.00	78.47 ± 0.06	59.38 ± 0.64	80.21 ± 0.00	

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9.0	27.08 ± 0.57	81.25 ± 0.58	69.44 ± 0.58	82.64 ± 0.46	66.67 ± 0.64	$84.38. \pm 0.00$
10.0	27.43 ± 0.00	85.07 ± 0.58	75.00 ± 0.63	85.76 ± 0.00	72.22 ± 0.11	88.89 ± 0.00
11.0	27.43 ± 0.10	88.19 ± 0.58	80.90 ± 0.00	90.28 ± 0.00	83.33 ± 0.00	92.71 ± 0.00
12.0	27.77 ± 0.05	89.58 ± 0.58	84.72 ± 0.00	92.71 ± 0.00	89.58 ± 0.00	96.53 ± 0.00

Table 6: Drug release kinetics models.

Vinatia madala	Formulations							
Kinetic models	F1	F2	F3	F6	F7			
		Zero-c	order					
K _o	0.259	0.193	0.375	0.170	0.380			
r ²	0.995	0.992	0.997	0.972	0.997			
		First-o	order					
K,	1.537	1.443	1.689	1.413	1.696			
r ²	0.862	0.942	0.985	0.974	0.986			
	'	Higu	ichi					
K _H	10.80	4.932	32.78	2.52	32.23			
r ²	0.997	0.978	0.975	0.924	0.974			
'	-	Hixson-C	Crowell	'				
K _{HC}	0.088	0.060	0.123	0.413	0.142			
r ²	0.995	0.977	0.861	0.892	0.953			
'	<u>'</u>	Korsmeye	r-peppas	'				
K _{KP}	0.426	0.482	0.270	0.449	0.282			
r ²	0.998	0.977	0.930	0.924	0.913			

lipid nanoparticles formulations were fitted into various kinetic models (Table 6). The findings revealed that zero-order model best-fitted the kinetic models that could be used to describe the release characteristics of olmesartan medoxomil loaded solid lipid nanoparticles formulations. In comparison with other kinetic models, the zero-order kinetics model possessed the highest correlation coefficients. Therefore, diffusion of the drug from the lipid matrix was suggested to be the probable mechanism of action. The values of release exponent (n) of all the formulations lies within n<0.5. These n values further revealed Fickian diffusion as the mechanism of action.

Conclusion

Olmesartan medoxomil was successfully incorporated into solid lipid nanoparticles by hot homogenization and ultrasonication method. The formulations were found to be in nanometric range with smooth spherical structure. The drug was found to be efficiently entrapped in the lipid matrix. FT-IR study revealed no interaction between drug and excipients. The release of drug from the formulations best-fitted Zero kinetics model equation and the probable mechanism of action indicates drug diffusion through the lipid matrix. Finally, the kinetics of *in-vitro* drug release results and other collected data suggest that solid lipid nanoparticles are potential delivery systems for olmesartan medoxomil.

References

- Lawrence MJ, Rees GD (2000) Microemulsions-based media as novel drug delivery systems. Advanced Drug Delivery Reviews 45: 89-12.
- Lee JM, Park KM, Lim SJ, Lee MK, Kim CK (2002) Microemlsion formulation of clonixic acid: solubility enhancement and pain reduction. I Pharm Pharmacol 54: 43-49.
- Ghosh PK, Murthy RS (2006) Microemulsions: A potential drug delivery system. Curr Drug Deliv 3: 167-180.
- 4. Baboota S, Shakeel F, Ahuja A, Ali J, Shafiq S (2007) Design development and evaluation of novel nanoemulsions formulations for transdermal

potential of celecoxib. Acta Pharm 8: 316-332.

- Shakeel F, Baboota S, Ahuja A, Ali J, Aqil M (2007) Nanoemulsions as vehicles for transdermal delivery of aceclofenac. AAPS Pharm Sci Tech 8: E104.
- Kotta S, Khan AW, Ansari SH, Sharma RK, Ali J. (2014). Anti HIV nanoemulsion formulation: optimization and *in-vitro-in-vivo* evaluation. Int J Pharm 462: 129-134.
- Allen TM, Moase EH (1996) Therapeutic opportunities for targeted liposomal drug delivery. Adv Drug Deliv Rev 21: 117-133.
- Ramos-Cabrer P, Campos F (2013) Liposomes and nanotechnology in drug development: focus on neurological targets. Int J Nanomedicine. 8: 951-960.
- 9. Jain S, Patil SR, Swarnakar NK, Agrawal AK. (2012) Oral delivery of doxorubicin using novel polyelectrolyte-stabilized liposomes (layersomes). Mol Pharm. 9: 2626-2635.
- 10. Ng HL, Lu A, Lin G, Qin L, Yang Z (2014) The potential of liposomes with carbonic anhydrase IX to deliver anticancer ingredients to cancer cells in vivo. Int J Mol Sci 16: 230-255.
- 11. Kshitija K, Swati M (2013) Self emulsifying drug delivery system: A review. Int J Pharm Chem Sci 2: 34-44.
- 12. Jingling T, Jin S, Fude C, Zhonggui H (2006) Preparation of self-emulsifying drug delivery systems of *Ginkgo biloba* extracts and *in-vitro* dissolution studies. Asian J Trad Med 1: 3-4.
- 13. Gajendra S, Khinchi MP, Gupta MK, Dilip A, Adil H, et al. (2012) Self emulsifying drug delivery systems (SEEDS): An approach for delivery of poorly water soluble drug. Int J Pharm Life Sci 3: 1991-1996.
- 14. Mehnert W, Mader K (2001) Solid lipid nanoparticles, production, characterization and applications. Adv Drug Deliv Rev 47: 165-196.
- Muller RH, Radtke M, Wissing SA (2002) Solid lipid nanoparticles (SLN) and nanostructured lipid carriers (NLC) in cosmetic and dermatological preparations. Adv Drug Deliv Rev 54: S131-S155.
- 16. Muller RH, Mader K, Gohla S (2000) Solid lipid nanoparticles (SLN) for controlled drug delivery-a review of the state of the art. Eur J Pharm Biopharm 50: 161-177.
- 17. Wissing SA, Kayser O, Muller RH (2004) Solid lipid nanoparticles for parenteral drug delivery. Adv Drug Deliv Rev 56: 1257-1272.
- 18. Bondi ML, Fontana G, Carlisi B, Giammona G (2003) Preparation and characterization of solid lipid nanoparticles containing cloricromene. Drug Delivery 10: 245-250.

- 19. Schubert MA, Muller-Goymann CC (2005) Characterisation of surfacemodified solid lipid nanoparticles (SLN). Influence of lecithin and nonionic emulsifier. Eur J Pharm Biopharm 61: 77-86.
- 20. Wong HL, Bendayan R, Rauth AM, Li YQ, Wu XY (2007) Chemotherapy with anti-cancer drugs encapsulated in solid lipid nanoparticles. Adv Drug Deliv Rev 59: 491-504.
- 21. Muller RH, Runge S, Ravelli V, Mehnert W, Thunemann AF, et al. (2006) Oral bioavailability of cyclosporine: solid lipid nanoparticles (SLN®) versus drug nanocrystals. Int J Pharm 317: 82-89.
- 22. Muller RH, Mehnert W, Lucks JS, Schwarz C, Zur Muhlen A, et al. (1995) Solid lipid nanoparticles (SLN)-An alternative colloidal carrier system for controlled drug delivery. Eur J Pharm Biopharm 41: 62-69.
- 23. Muller RH, Mader K, Gohla S (2000) Solid lipid nanoparticles (SLN) for controlled drug delivery-a review of the state of the art. Eur J Pharm Biopharm 50: 161-177.
- 24. Lee J, Kim A, Kyung-Sang Yu K, Jae-Yong Chung J, et al. (2015) Pharmacokinetics and bioequivalence of two different 20 mg olmesartan tablets: A randomized, single-dose, two-period crossover study in healthy Korean male volunteers. Trend Clin Pharmacol 23: 49-53.
- 25. Warne, GT, Jarvis B (2002) Olmesartan medoxomil. Drugs 62: 1345-1353.
- 26. Thomson PDR (2005) Physcian desk reference. New Jersey.
- 27. Brunner HR (2002) The new oral angiotensin II antagonist olmesartan medoxomil: a concise overview. Journal of Human Hypertension 16: \$13-\$16.
- 28.Thakkar HP, Patel BV, Thakkar SP (2011) Development and characterization of nanosuspension of olmesartan medoxomil for bioavailability enhancement. J Pharm Bio Allied Sci 3: 426-434.

- 29. Sasidhar RLC, Vidyadhara S, Maheswart GV, Deepti B, Babu S (2013) Solubility and dissolution rate enhancement of olmesartan medoxomil by complexation and development of mouth dissolving tablets. Adv Biol Res 7: 32-41.
- 30. Raj VP, Debnath S, Maleswari T, Babu MN, Naik BB (2013) Preparation, characterization and evaluation of olmesartan medoxomil-β cyclodextrin complexes. Ind J Res Pharm Biotech 1: 420-427.
- 31. Sathali AA, Jayalakshmi J (2013) Enhancement of solubility and dissolution rate of olmesartan medoxomil by solid dispersion technique. J Curr Chem Pharm Sci 3: 125-134.
- 32. Yadav AA, Yadav DS, Karekar PS, Pore YV, Gajara P (2012) Enhanced solubility and dissolution rate of olmesartan medoxomil using crystalloco-agglomeration technique. Der Pharmacia Sinica 3: 160-169.
- 33. Zhang H, Yu L (2004) Dissolution Testing for Solid Oral Drug Products: Theoretical Considerations. American Pharmaceutical Review 7: 26-31.
- 34. Tadey T, Carr G (2009) Dissolution Testing for Solid Oral Dosage Forms. Pharmaceutical Formulation and Quality 11: 22-27.
- 35.Suvakanta D (2010) Kinetic modeling on drug release from controlled drug delivery system. Acta Polonia Pharmaceutical-drug research 67: 217-223.
- 36. Hussain L (2013) Kinetic modeling and dissolution profiles comparison: An overview. Int J Pharm Bio Sci 4: 728-737.
- 37. Ekambaram P, Sathali AH, Priyanka K (2012) Solid Lipid Nanoparticles: A Review. Scientific Reviews Chemical Communications 2: 80-102.
- Charman WN, Stella VJ (1992) Lymphatic transport of drugs. CRC Press, Boca Raton.
- 39. Zhang J, Fan Y, Smith E (2009) Experimental design for the optimization of lipid nanoparticles. J Pharm Sci 98: 1813-1819.

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